



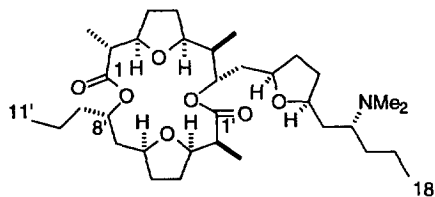
The Total Synthesis of Pamamycin 607. 1. Synthesis of a C1'-C11' Synthon.

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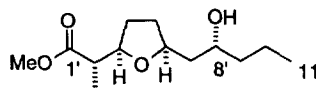
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Abstract: The synthesis of a C1'-C11' synthon **2** of pamamycin-607 starting from alcohol **3** in ten steps is reported. Copyright © 1996 Elsevier Science Ltd

The pamamycins form a group of homologous, naturally-occurring macrodiolides.¹ They possess a remarkable range of chemical and biological activity, including antibiotic properties and anionophoric behaviour.²⁻⁵ In our planned synthesis of the homologue of molecular weight 607 (**1**), we have identified the C8'-epimer of the C1'-C11' subunit as a useful intermediate (synthon **2**). Our intention is to couple this with the C1-C18 subunit using a process which would involve inversion at C8'.⁶ Should this process prove difficult, inversion⁷ of C8' prior to macrolactonisation⁸ of the subunits could be employed.

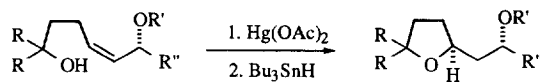


1 Pamamycin 607



2 C1'-C11' synthon

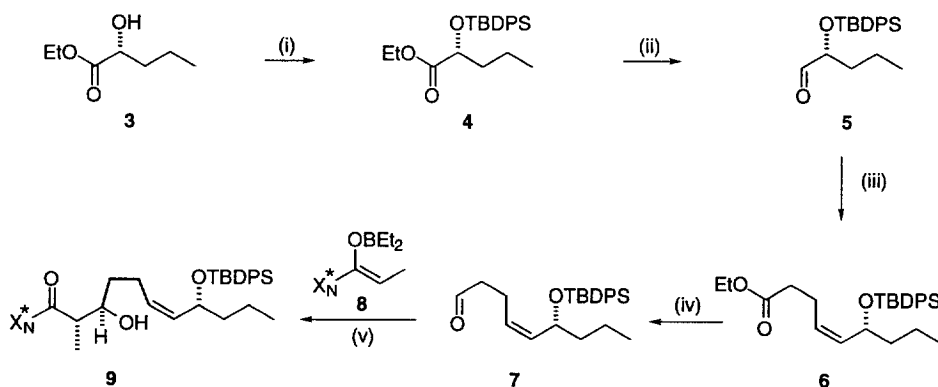
We have recently carried out studies^{9,10} on the diastereoselectivity associated with the ring closure of γ -hydroxyalkenes bearing a remote, secondary allylic ether to give substituted tetrahydrofurans (Scheme 1). We found good to excellent diastereoselectivity ranging from 6:1 up to 20:1. Herein we report an extension of this methodology to the synthesis of the C1'-C11' synthon (**2**).



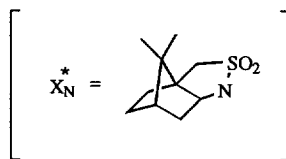
Scheme 1

General scheme for intramolecular oxymercuration/demercurations.

The synthesis of the ring closure precursor **9** was accomplished in five steps from the known ethyl (2*S*)-2-hydroxypentanoate **3** (ee. 85-89%)^{11,12} as outlined in Scheme 2. Protection of the secondary alcohol as its *tert*-butyldiphenylsilyl (TBDPS) ether (96%), followed by diisobutylaluminum hydride (DIBAL) reduction (95%) gave the aldehyde **5**. *Z*-Selective olefination (51%) provided alkene **6** with good selectivity (*Z*:*E* = 89:11). A second DIBAL reduction proceeded cleanly to give the new aldehyde **7** in 97% yield. This alkenal was then subjected to a *syn*-selective aldol reaction using the diethylboron enolate **8**¹³ of Oppolzer's camphor-derived *N*-propionylsultam to give the required major (*syn*) product **9** (54%).¹⁴



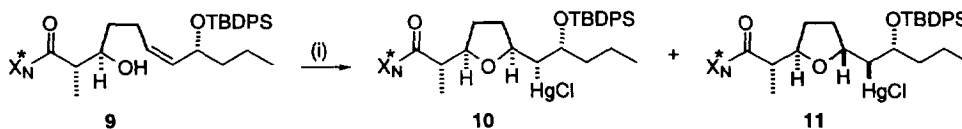
- (i) TBDPSCl, Im., DMF, DMAP, r.t., 12h.
 (ii) DIBAL, toluene, -78°C, 2h.
 (iii) Ph₃P=CHCH₂CH₂CO₂Et, THF, 0°C.
 (iv) DIBAL, toluene, -78°C, 1h.
 (v) CH₂Cl₂, -78°C, 75min.



Scheme 2 Synthesis of ring closure precursor **9**.

Ring closure was then effected using the intramolecular oxymercuration procedure outlined earlier.^{9,10} Thus, when an acetonitrile solution of alkenol **9** was treated with 1.5 equivalents of mercury(II)acetate for 18h at room temperature the products, isolated as their chloromercurial derivatives, were obtained in a 6:1 ratio in

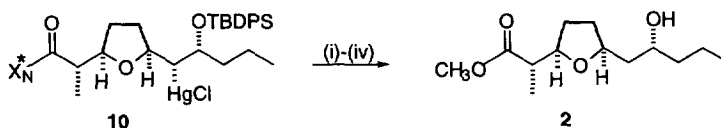
favour of the desired diastereomer **10** (Scheme 3). This molecule, which could be easily separated from **11** by chromatography on silica gel, contains the complete skeleton of the target synthon (**2**) with all the stereocentres correctly installed.



(i) (a) $\text{Hg}(\text{OAc})_2$, CH_3CN , r.t., 18h; (b) Aq. NaCl .

Scheme 3 Intramolecular oxymercuration of alkenol **9**.

Four straightforward functional group manipulations remained to complete the synthesis of **2**. Reductive demercuration with tributylstannane and AIBN¹⁵ proceeded in a 93% yield. Removal of the chiral auxiliary was achieved by hydrolysis with alkaline hydrogen peroxide¹³ and the crude reaction mixture was directly esterified with an excess of diazomethane in 46% yield over 2 steps. Finally, desilylation with tetrabutylammonium fluoride produced the target molecule **2** in 56% yield $[[\alpha]_D^{20} -6.0^\circ (c\ 0.5, \text{CHCl}_3)]^{16}$ (Scheme 4).



(i) Bu_3SnH , AIBN, toluene; (ii) 30% Aq. H_2O_2 , LiOH , THF; (iii) CH_2N_2 , Et_2O ; (iv) TBAF, THF, r.t.

Scheme 4 Completion of the synthesis of **2**.

The application of this approach to the synthesis of the larger subunit¹⁷ of pamamycin 607 is under way in our laboratories.

ACKNOWLEDGEMENTS

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- 14 Minor impurities (due to the incomplete enantioselectivity in the reduction leading to hydroxyester **3**) were difficult to separate and were carried through to compound **2** where they could be completely removed.
- 15 Evans, D. A.; Bender, S. L.; Morris, J. *J. Am. Chem. Soc.* **1988**, *110*, 2506-2526.
- 16 Data for **2**: $[\alpha]_D^{20}$ -6.0° (c 0.5, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 0.92 (t, *J*=7.0 Hz, 3H, CH₃), 1.22 (d, *J*=7.0 Hz, CH₃), 1.31-1.39 (m, 2H, CH₂), 1.41-1.51 (m, 4H, CH₂x2), 1.60-1.74 (m, 2H, CH₂), 1.91-2.07 (m, 2H, CH₂), 2.57 (p, *J*=7.1 Hz, 1H, CHCH₃), 3.54 (bs, 1H, OH), 3.69 (s, 3H, OCH₃), 3.70-3.83 (m, 1H, CHOTBDPS), 3.99-4.06 (m, 2H, CHOx2). ¹³C NMR (50 MHz) δ 174.8 (C=O); 80.9 (CHO), 80.4 (CHO), 71.5 (CHOH), 51.7 (OCH₃); 44.8 (CHCH₃); 42.7 (CH₂CO); 39.7 (CH₂CO); 31.9 (CH₂COH); 28.6 (CH₂); 18.7 (CH₂); 14.1 (CH₃CH); 13.9 (CH₃). MS (CI): 245 (M+1, 32%), 228(20), 227(100), 209(20), 195(40), 157(80), 139(28), 71(15), 57(20). HRMS (CI) Calcd. for C₁₃H₂₄O₄: 245.176. Found 245.176 ± 0.002.
- 17 For an alternative approach to an intermediate for this subunit, see: Walkup, R. D.; Kim, Y. S. *Tetrahedron Lett.* **1995**, *36*, 3091-3094.

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